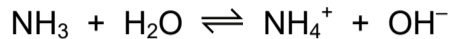


19 Aqueous ammonia forms an equilibrium with water.



In this equilibrium, which particle acts as an acid and which particle acts as a base?

	acid	base
<b>A</b>	$\text{NH}_3$	$\text{H}_2\text{O}$
<b>B</b>	$\text{NH}_3$	$\text{OH}^-$
<b>C</b>	$\text{H}_2\text{O}$	$\text{NH}_3$
<b>D</b>	$\text{H}_2\text{O}$	$\text{NH}_4^+$