

**21** Phosphorus is an element in Group V of the Periodic Table.

It burns in air to form an oxide that dissolves in water to form a solution with a pH of 1.

Which row describes this oxide of phosphorus?

	metal oxide	non-metal oxide	acidic oxide	basic oxide
<b>A</b>	✓	✗	✓	✗
<b>B</b>	✓	✗	✗	✓
<b>C</b>	✗	✓	✓	✗
<b>D</b>	✗	✓	✗	✓

key

✓ = yes

✗ = no