

14 A sealed container of gas is heated. The pressure of the gas increases.

Which statement explains this increase in pressure?

- A** The forces of the gas particles striking the walls of the container increase.
- B** The forces of attraction between the gas particles and the walls of the container increase.
- C** The gas particles collide with each other more frequently.
- D** The gas particles lose more energy when they strike the walls of the container.